

1 **Acute toxicity of zinc and arsenic to the warmwater aquatic oligochaete**
2 ***Branchiura sowerbyi* as compared to its coldwater counterpart *Tubifex tubifex***
3 **(Annelida, Clitellata)**

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26 **Abstract**

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28 *Purpose* This study aimed at evaluating the acute effects of arsenic and zinc to the warmwater
29 aquatic oligochaete *Branchiura sowerbyi*. Relative sensitivity with the coldwater species
30 *Tubifex tubifex* was compared. Implications for the use of *B. sowerbyi* in the risk assessment of
31 sediments in the tropics are discussed.

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Materials and methods Water-only (96h) and sediment (14d) toxicity tests were conducted with
both species evaluating a concentration series of arsenic and zinc. The tests were conducted
considering the environmental conditions in the natural habitat of *T. tubifex* (predominantly
temperate) and *B. sowerbyi* (predominantly tropical). Both lethal and sublethal endpoints
(autotomy of the posterior body parts, abnormal behavior and appearance) were determined in
the tests. The lethal (LC₁₀ and LC₅₀) and effect (EC₁₀ and EC₅₀) concentrations were also
determined to assess metal sensitivity for both species.

Results and discussion Both test species were more sensitive to Zn than As in water-only tests,
which is in agreement with previous studies evaluating the toxicity of these metals to aquatic
oligochaetes. Sublethal effects were generally noted at concentrations lower than those leading
to mortality. The warmwater oligochaete *B. sowerbyi* was more sensitive to both metals tested
than the coldwater species *T. tubifex*.

Conclusions Study findings support the need for using indigenous tropical species in risk
assessments in the tropics. In addition, sublethal effect parameters should be included in toxicity
testing with aquatic oligochaetes.

Keywords Autotomy • Bioassay • Ecotoxicology • Survival • Tropics • Tubificines

49 **1 Introduction**

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3 50 The tropical zone **contains** the highest species diversity on the planet. On the other
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5 51 hand, many countries in the tropics are developing nations, heavily populated, and
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7 52 rapidly becoming industrialized but lack the money, infrastructure, and other resources
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9 53 for advanced pollution controls (Kwok et al. 2007). Subsequently, aquatic risk
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11 54 assessments in tropical countries often rely on studies from temperate countries, even
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13 55 though the fate and effects of **contaminants** may be different between climatically
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15 56 distinct regions (Daam and Van den Brink 2010). The use of temperate toxicity data in
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17 57 tropical risk assessments has been disputed (e.g. Lahr et al. 2001; Do Hong et al. 2004;
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19 58 Lopes et al. 2007; Freitas and Rocha 2011).

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22 59 Oligochaetes feed on bulk sediment and penetrate the sediment through the construction
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24 60 of burrows that extent into anoxic sediments. These traits make them especially
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26 61 vulnerable to sediment contamination (Warren et al. 1998). By bioturbation of the
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28 62 sediment and by serving as prey these animals can have a strong influence on the
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30 63 bioavailability of contaminants to other organisms, e.g. benthivorous fish (OECD
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32 64 2007).

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35 65 **The** oligochaete *Tubifex tubifex* (Müller 1774) (Clitellata, Tubificinae) is widely
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37 66 available and abundant in temperate world regions, easy to keep in the laboratory and
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39 67 can be easily cultured under laboratory conditions. Subsequently, this species has often
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41 68 been used in ecotoxicology testing and test protocols have been developed (e.g. ASTM
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43 69 2005; OECD 2008). *T. tubifex* is most commonly encountered in temperate regions and
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45 70 is associated with cold water in tropical regions (Bonacina et al. 1987).

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48 71 **B***ranchiura sowerbyi* (Beddard 1892) (Clitellata, Rhyacodrilinae) has only more
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50 72 recently been indicated to be an adequate alternative aquatic oligochaete test species for
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52 73 ecotoxicology testing (e.g. Marchese and Brinkhurst 1996; Ducrot et al. 2007; Saha and
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74 Kaviraj 2008; Ducrot et al. 2010; Del Piero et al. 2014; Lobo and Espíndola 2014). *B.*
75 *sowerbyi* appears to prefer warm waters and in temperate latitudes is often found in
76 artificially heated habitats, although it is not restricted to them (Bonacina et al. 1994).
77 The most common endpoint in acute toxicity testing with oligochaetes is mortality (e.g.
78 ASTM 2005; OECD 2007). Another reaction of oligochaetes to metal stress is autotomy
79 (“self-amputation”) of the posterior body parts, which aids the worms in escaping from
80 predators as well as in body detoxification (Fleming et al. 2007). Oligochaetes
81 accumulate metals in their posterior body segments, which can be discarded when the
82 concentration reaches critical values (Lucan-Bouché et al. 1999; Rathore and Khangarot
83 2003).
84 Arsenic may be present in the aquatic environment from natural and anthropogenic
85 sources. In Brazil, arsenic contamination has been encountered in mining regions, with
86 sediment concentrations of up to 3300 mg kg⁻¹ d.w. (Deschamps et al. 2002). Zinc has
87 also been detected in high concentrations in the aquatic environment. For example,
88 concentrations up to 574 mg Zn kg⁻¹ d.w. were reported in sediment from the Tietê river
89 basin (São Paulo State, Southeast region of Brazil), a concentration that is 80 times
90 greater than the global geological reference level of 6 mg kg⁻¹ d.w. (Nascimento and
91 Mozeto 2008). Despite the frequently detected contamination of sediments with arsenic
92 and zinc, little is presently known about the (sublethal) effects of these substances to
93 aquatic oligochaetes, especially in tropical regions.
94 The aim of the present study was to study the lethal and sublethal (autotomy of the
95 posterior body parts, abnormal behavior and appearance) effects of arsenic and zinc on
96 the warmwater oligochaete *B. sowerbyi*. Differences in sensitivity with the coldwater
97 oligochaete *T. tubifex* were determined by also conducting acute tests with the latter
98 species. Tests for both species were conducted in the presence and absence of sediment

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99 with the objective to determine the sensitivity differences of the test organisms to water
100 and sediment exposures.

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102 **2 Materials and methods**

103 The tests with *T. tubifex* and *B. sowerbyi* were conducted at the Animal Ecotoxicology
104 and Biodiversity Laboratory of the University of the Basque Country (UPV/EHU,
105 Spain) and Nucleus for Ecotoxicology and Applied Ecology (NEEA) of the University
106 of São Paulo (Brazil), respectively. The tests were conducted considering the
107 environmental conditions in the natural habitat of *T. tubifex* (predominantly temperate)
108 and *B. sowerbyi* (predominantly tropical).

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110 **2.1 Culture and maintenance of test organisms**

111 The test organisms were obtained from existing in-house cultures of the laboratories
112 where the tests were conducted. The *T. tubifex* culture has been maintained in the
113 Animal Ecotoxicology and Biodiversity Laboratory for over 20 years. Animals are kept
114 at 22 ± 1 °C in complete darkness in vessels (length: 15 cm; width: 15 cm; height: 10
115 cm) containing a 3-cm sediment layer (grain size < 0.25 mm) obtained from an
116 **uncontaminated** pond in the mountains of Álava (Iturbatz, Spain) supplied with
117 groundwater. The 3 cm water column is made up of dechlorinated tapwater (pH = $6.8 \pm$
118 0.2 ; electrical conductivity (EC) = 279 ± 3 $\mu\text{S cm}^{-1}$; total hardness = 127 mg $\text{CaCO}_3 \text{ L}^{-1}$)
119 and is moderately aerated. To ensure that tests are performed with similarly aged
120 animals, fresh cultures are initiated at six to seven weeks (i.e. the time for animals to
121 reach maturity) before testing by transferring 100 to 150 small juvenile animals from
122 the existing stock culture to a new vessel. A sediment with poorer organic content from

123 Barrundia (Álava, Spain) was sampled for the performance of the toxicity tests with *T.*
124 *tubifex*.
125 *B. sowerbyi* has been maintained at NEEA for approximately four years under $25 \pm 1^\circ\text{C}$
126 in the dark with moderate aeration. Each 3.5-L vessel contains a sediment layer (grain
127 size $< 0.50\text{mm}$) of approximately 5 cm derived from the waterbody (Perdizes stream)
128 alongside a spring in Brotas (São Paulo State, Brazil). Reconstituted test water was used
129 with the following physical-chemical characteristics: $\text{pH} = 7.0$; $\text{EC} = 100 \mu\text{S cm}^{-1}$; total
130 hardness = 40 mg L^{-1} (as CaCO_3). Test organisms were acclimated by transferring 80
131 small juveniles to a newly started culture six to nine weeks prior to the start of the tests,
132 which is the maturity time of this species.
133 In the field, 0.5 kg (wet weight) sediment was collected with a stainless steel spade and
134 sieved through 0.5 mm to remove associated macrofauna, which may influence the
135 subsequent analytical results (Reynoldson 1994). The sediment was assessed for particle
136 size distribution (by sieving), total organic matter content (calculated through the
137 percent loss of organic matter by ignition at 450°C for 6h), water content (105°C
138 overnight), and metal concentrations (using ICP-AES in Spain and AAS in Brazil).
139 Results from sediment analyses as used in the test cultures and toxicity tests are
140 presented in Table 1.

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142 **2.2 Water-only exposure (96h)**

143 The 96h water-only tests with both *B. sowerbyi* and *T. tubifex* followed the method as
144 described by Maestre et al. (2009). Each test included the exposure of worms to each
145 metals separately, and consisted of a control and five metal concentrations ranging for
146 arsenic (spiked as $\text{HAsNa}_2\text{O}_4 \cdot 7\text{H}_2\text{O}$; purity 98%) from 0 to 81 mg As L^{-1} in *B. sowerbyi*
147 and from 0.03 to 118 mg As L^{-1} in *T. tubifex* exposure; and for zinc (spiked as

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148 ZnSO₄·7H₂O; purity: 99%) from 0 to 8.9 mg Zn L⁻¹ in *B. sowerbyi* and 0.03 to 41 mg
149 Zn L⁻¹ in *T. tubifex* exposure (Table 2). The tested concentrations were chosen based on
150 a literature review and on preliminary experiments conducted in the laboratory
151 (unpublished data). Tests were conducted in 250 mL beakers (pre-washed with 10%
152 nitric acid) with 100 mL test solution, five replicates per concentration and five worms
153 per replicate. The test lasted for 96h, in the dark and with neither food nor aeration.
154 Nominal and measured concentrations tested are provided in Table 2. The test
155 organisms from *B. sowerbyi* cultures were 7-8 weeks old (sexually mature with visible
156 eggs in the ovisac), and 6-8 weeks old in the case of *T. tubifex* (sexually mature with a
157 well-developed clitellum). At the beginning of the test, a 10 mL water sample was taken
158 from each metal concentration of the treatments, acidified with 70% nitric acid (150 µL
159 for each 100 mL sample) and stored at 4°C until chemical analyses (see Section 2.4).
160 The number of dead animals and any visible sublethal effects (i.e. abnormal behavior
161 and appearance) were noted daily. A worm was considered to be dead when there was
162 no response in the 10 s after receiving a slight disturbance with a bar (following Maestre
163 et al. 2009 and references therein). Dead animals were removed to avoid any adverse
164 influence on physical-chemical characteristics resulting from decomposition.
165 Autotomy, i.e. self-amputation of segments through a local macroscopic constriction of
166 the circular muscles of the caudal region of a worm has frequently been described as a
167 response to toxic stress induced by metals (Maestre et al. 2009 and references therein).
168 Therefore, the number of organisms that suffered autotomy was also quantified at the
169 end of the 96h experimental period.

171 2.3 Sediment toxicity test (14d)

172 The sediment toxicity tests were conducted with arsenic and zinc for *T. tubifex* and with
173 Zn for *B. sowerbyi*. For logistic reasons, the test with *B. sowerbyi* evaluating arsenic
174 could unfortunately not be conducted.

175 Test sediments were spiked with metal solutions to achieve five concentrations,
176 following the recommendations of OECD (2007). The sediment volume required to
177 conduct the tests and the quantity of the test compound to obtain the desired nominal
178 concentrations were mixed in a 5-L vessel with dilution water in a ratio of 1:4 (sediment
179 weight: water volume). Tetramin[®] (80 grams/100 mL sediment) was added after which
180 the vessels were shaken for four hours at moderate speed (160-170 rpm) on an orbital
181 shaker (KS501D, IKA Labortechnik GmbH, Staufen, Germany). Subsequently, vessels
182 were maintained for one week at test temperature (22 ± 1 °C for *T. tubifex* and 25 ± 1 °C
183 for *B. sowerbyi*) to allow sediment and porewater partitioning equilibrium. After this
184 period the overlying water was carefully removed through siphoning, mechanically
185 homogenized and distributed over the test containers. Each test container received 100
186 mL sediment and 100 mL fresh dilution water and was kept at their respective test
187 temperature for 48h under moderate aeration. Then test animals were added at which
188 time the experimental observation period commenced.

189 Toxicity tests were conducted, with modifications, according to ASTM (2005). To this
190 end, three (*T. tubifex*) or five (*B. sowerbyi*) replicates were prepared for each five test
191 concentration as well as for the untreated controls, where one replicate for chemical
192 analysis at the start of the toxicity tests, and the others replicates were used for effect
193 assessments. Test concentrations ranged from 12 to 41 mg As kg⁻¹ d.w. and 181 to 679
194 mg Zn kg⁻¹ d.w. for *T. tubifex* exposure, and from 181 to 679 mg Zn kg⁻¹ d.w. for *B.*
195 *sowerbyi* exposure (Table 2). Each replicate contained four test organisms in their start
196 of the first reproductive cycle (6-7 weeks for *T. tubifex* and 7-8 weeks for *B. sowerbyi*),

197 all obtained from the same cohort. At the end of the experimental period (i.e. 14 days
198 after introducing the test organisms), the sediments were washed over a 0.5 mesh sieve
199 and the collected organisms assessed for mortality and autotomy.

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201 **2.4 Physical-chemical water quality parameters**

202 In all the water-only and sediment toxicity tests, the following parameters were
203 measured daily: DO (Yellow Springs YSI-55, Ohio, USA), pH (Micronal B374, São
204 Paulo, Brazil), EC (Thermo Orion M145, Beverly, USA) and ammonia
205 (spectrophotometry method, APHA 1995) in the *B. sowerbyi* test. DO, pH and EC were
206 measured with a Thermo Orion 5-Star Plus (Beverly, USA) meter in *T. tubifex* toxicity
207 tests.

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209 **2.5 Metal analyses**

210 Metal concentrations in the water-only and sediment tests were measured as described
211 below. In the latter, the whole-sediment and the porewater concentrations were analyzed
212 from the replicate used for chemical analyses. The porewater was removed from the
213 sediment through centrifugation of 50 mL sediment (3000 rpm for 30 min at 4°C), and
214 subsequently filtered through a 0.45 µm filter before chemical analyses (Méndez-
215 Fernández et al. 2013). The whole-sediment sample was dried at ambient temperature,
216 sieved through a 0.063 mm mesh before digestion and analyzed as outlined below.

217 Arsenic and zinc concentrations in water from *T. tubifex* tests were performed by
218 SGIker at the University of Basque Country using Inductively Coupled Plasma-Atomic
219 Emission Spectroscopy (ICP-AES) or ICP-Mass spectroscopy (ICP-MS), depending on
220 metal concentrations. Detection limits (dl) were 0.05 and 0.10 mg L⁻¹ for ICP-AES and
221 5 and 0.3 µg L⁻¹ for ICP-MS, for arsenic and zinc respectively, and analytical recoveries

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222 were 113% for arsenic (NIST 1643e) and 96-104% for zinc (TMDA 52.3 - NIST
223 1643e). Sediment metal concentration was measured by Analytical Service of
224 Sosprocan Unit (University of Cantabria, Santander, Spain), following the EPA 3052
225 protocol for microwave assisted acid digestion of siliceous- and organically-based
226 matrices and the UNE-EN 13656:2003-03-28 method for microwave assisted digestion
227 with hydrofluoric (HF), nitric (HNO₃) and hydrochloric (HCl) acid mixture for
228 subsequent determination of elements. After digestion, metal concentrations were
229 determined using ICP-MS (dl = 5 $\mu\text{g L}^{-1}$ for Zn and 0.3 $\mu\text{g L}^{-1}$ for As). Buffalo River
230 Sediment reference material (RM8704, NIST USA) was also analyzed for quality
231 control and recoveries were within certified values (mean of 87% for zinc).
232 Analytical quantification of the test compounds in water and sediment from the *B.*
233 *sowerbyi* tests was performed in the Poços de Caldas laboratory (LAPOC) of the
234 National Nuclear Energy Commission (CNEN) in Brazil. Water samples were analyzed
235 by flame atomic absorption spectroscopy (F-AAS; dl = 20 $\mu\text{g L}^{-1}$) for zinc and by
236 hydride generation atomic absorption spectrometry (HG-AAS; dl = 6 $\mu\text{g L}^{-1}$) for arsenic.
237 Reference water (NIST 1643e) yielded an analytical recovery of 103% and 107% for
238 zinc and arsenic, respectively. Sediment was digested following EPA 3052, and
239 analyzed using F-AAS (for zinc) and HG-AAS (for arsenic), as described above. The
240 reference sediment of Buffalo River yielded a recovery rate of 90% for zinc.

241 242 **2.6 Statistical analyses**

243 Statistical analyses were conducted using the measured test concentrations. Two-
244 parameter log-logistic (LL.2) regression **models** were used to calculate the Effective and
245 Lethal Concentrations (EC₁₀, EC₅₀, LC₁₀ and LC₅₀). Significant differences with
246 controls were assessed by Fisher's exact test. Significance level was $\alpha = 0.05$. Statistical

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247 analyses were performed using the free R software (R core team 2013) and extension
248 package *drc* (Ritz and Streiberg 2005).

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250 **3 Results and discussion**

251 **3.1 Test performance**

252 The physical-chemical conditions were comparable between the different experiments
253 and varied little between treatments as indicated by the low standard deviation values
254 (Table 3). Conditions were within the ranges as indicated in existing standard protocols
255 for oligochaete testing, e.g. pH between 6 and 9 (OECD 2007, 2008) and dissolved
256 oxygen concentration above 2.5 mg L⁻¹ in the overlying water (ASTM 2005). Ammonia
257 concentrations remained below the levels considered toxic to oligochaetes (5 mg L⁻¹;
258 Schubauer-Berigan et al. 1995; see Table 3).

259 Appropriate test conditions were also confirmed by the control treatment performances.
260 In the water-only test with *T. tubifex*, one test organism had died (which equals 5% of
261 the total test population) and another one showed autotomy of the posterior body region
262 (5%). No other lethal or sublethal effects on control organisms were recorded.

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264 **3.2 Effects of arsenic and zinc on oligochaete survival**

265 The effects of zinc and arsenic on survival and autotomy of posterior body section
266 observed in the water-only tests with *T. tubifex* and *B. sowerbyi* are illustrated in Fig. 1.
267 The effective concentrations calculated for all water-only as well as the sediment tests
268 are summarized in Table 4.

269 To the authors' knowledge, this study is the first to evaluate the toxicity of arsenic to *B.*
270 *sowerbyi*, whereas two water-only studies are available for *T. tubifex*. These studies

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271 reported a 96h-LC₅₀ of 8.9 mg As L⁻¹ (Khangarot 1991) and 127 mg As L⁻¹ (Fargašová
272 1994). The results of the present study are in line with the latter study (28% mortality at
273 the highest test concentration: 118 mg As L⁻¹). The greater toxicity as noted in the study
274 by Khangarot (1991) may be related to one or more of the following factors in the
275 experimental design of that study: i) test water was renewed every 24h; ii) possible
276 greater sensitivity of the test organism culture; iii) the metal salt tested in **the** present
277 study (HAsNa₂O₄·7H₂O) was different from the one used by Fargašová (1994)
278 (Na₃AsO₃); and iv) test temperature was 30°C, instead of 25°C and 22°C in Fargašová
279 (1994) and present study, respectively. The last factor has indeed been demonstrated to
280 influence the toxicity of metals, and several studies have reported greater toxicity with
281 increasing temperatures (Wang 1987; Rathore and Khangarot 2002).

282 Regarding the effects of zinc on survival, the 96h-LC₅₀ calculated in the water-only test
283 with *T. tubifex* of the present study (8.7 mg L⁻¹) was similar to that reported in a study
284 by Rathore and Khangarot (2002; 8.9 mg L⁻¹). *B. sowerbyi*, however, was more
285 sensitive in the sediment test (EC₁₀: 269.11 ± 46.14 mg kg⁻¹ d.w.) as compared with a
286 study conducted by Ducrot et al. (2010). The latter study only demonstrated a
287 significant decrease in *B. sowerbyi* juvenile survival rates in worms exposed to zinc
288 concentrations exceeding 1819 mg kg⁻¹ d.w. for 14, 21 and 28 days. Differences in the
289 adopted experimental design (e.g. temperature and pH of water, characteristics of
290 natural sediment used and sediment equilibration period) may be partly responsible for
291 this difference in observed effects. For example, Lee et al. (2004) demonstrated that
292 LC₅₀ values for zinc obtained for the amphipod *Leptocheirus plumulosus* were twice as
293 high in sediment-water tests with an equilibration period of 20 days as compared to tests
294 allowing only a 5-day equilibration period after sediment spiking. In the present study,

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295 an equilibration period of 10 days was used, whereas Ducrot et al. (2010) only allowed
296 the sediment to equilibrate for 3 days.

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298 **3.3 Sublethal effects on autotomy and behavior**

299 Autotomy was noted in the water-only and sediment experiments of both species and
300 metals (Fig. 2; Table 4). Other studies have also reported autotomy in oligochaetes as a
301 response and elimination mechanism to metal stress (e.g. Lucan-Bouché et al. 1999;
302 Rathore and Khangarot 2003; Méndez-Fernández et al. 2013). For example, EC₅₀ values
303 reported for autotomy by Méndez-Fernández et al. (2013) were five to ten times lower
304 than the LC₅₀ values for the three metals (Cd, Cu and Cr) tested on *T. tubifex*. In the
305 present study, however, little difference in toxicity levels for autotomy and mortality
306 was found after short-term water-only exposure: autotomy occurred at the same
307 concentrations as mortality in most cases. Only for intermediate Zn concentrations also
308 surviving organisms showed autotomy (Fig.1b, d). This is also reflected in the EC₅₀ and
309 EC₁₀ values, where no differences were found between mortality and autotomy for both
310 species (Table 4). Regarding autotomy in sediment tests after short-term exposures,
311 autotomy was noted at lower concentrations than mortality for both metals in *T. tubifex*
312 exposure, although the **standard error (SE)** of the effective concentration indicate that
313 the 95% confidence intervals overlapped (Table 4).

314 Abnormal appearance and behavior of the test organisms were observed at
315 concentrations below those leading to mortality or autotomy (Fig. 2). For example,
316 swelling of the body of *T. tubifex* was noted for 13% and 43% of the worms exposed for
317 24 hours to concentrations above 2.75 mg Zn L⁻¹ and 7.98 mg As L⁻¹, respectively (Fig.
318 2a). Swelling was accompanied by an increase in the number of granules in the
319 coelomic cavity (Fig. 2b). After 96 hours exposure to 118 mg As L⁻¹, one *T. tubifex*

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320 individual showed a deformed posterior body (Fig. 2c). In the tests with *B. sowerbyi*,
321 swelling of the body, posterior part lacking gills and with ring constrictions were
322 observed (Fig. 2d and 2e). Similar deformations have previously been demonstrated in
323 oligochaetes exposed to insecticides (Komala 1992), cadmium (Bailey and Liu 1980)
324 and okadaic acid (Franchini and Marchetti 2006).

325 At the end of the 14d exposure period in the sediment tests, five *T. tubifex* had a swollen
326 body cavity after exposure to arsenic concentrations greater than 208 mg As kg⁻¹ d.w.
327 No other effects on general appearance were recorded. The presence of sediment has
328 frequently been reported to potentially decrease toxicity through binding of the toxicant
329 to the sediment (e.g. Chapman et al. 1982). On the other hand, endobenthic aquatic
330 oligochaetes ingest sediment particles which may hence be a relevant uptake route and
331 subsequently increase toxicity. In the higher Zn (> 358 mg Zn kg⁻¹ d.w. for *B. sowerbyi*
332 and 679 mg Zn kg⁻¹ d.w. for *T. tubifex*) and As (421 mg As kg⁻¹ d.w. for *T. tubifex*)
333 treatments, however, organisms were noted to remain on the sediment surface and hence
334 avoiding penetrating the sediment, which may have reduced the toxicity due to a lesser
335 sediment intake and contact. However, this behavioral effect as well as those on general
336 appearance described above are likely to increase predation in natural environments,
337 and thus, the relevance and importance of such behavior needs further attention in future
338 studies.

339 340 **4 Conclusions and implications of study findings for the use of *B. sowerbyi* in** 341 **tropical aquatic risk assessments**

342 In the water-only toxicity tests with both species, toxicity values as calculated for zinc
343 were lower than those established for arsenic (Table 4). This is in agreement with
344 previous studies into the sensitivity of oligochaetes to metals that report a relatively

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345 high toxicity of zinc and a relatively low toxicity of arsenic among the metal tested (e.g.
346 Fargašová 1994; Rathore and Khangarot 2002). Effects on behavior and appearance of
347 the test organisms were noted at concentrations lower than those leading to significant
348 mortality levels. Subsequently, such parameters should be evaluated with care when
349 conducting aquatic risk assessment studies.

350 The warmwater oligochaete *B. sowerbyi* was more sensitive to both metals tested than
351 the coldwater species *T. tubifex*. This may be partly related to differences in
352 experimental set-up of the tests with these species (e.g. temperature, water chemistry
353 and physical-chemical characteristics of the sediment). In the light of these results, and
354 in accordance with previous authors (e.g. Daam and Van den Brink 2010 and reference
355 therein), we recommend the use of tropical species in the risk assessment of tropical
356 regions. There is thus also a need for the development of adequate protocols for
357 sediment toxicity assessment in these areas.

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359

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374 **Conflict of Interest:**

375 The authors declare that they have no conflict of interest.

376

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484 **Figure captions**

485

486 **Fig. 1.** Percentage mortality (black columns) and autotomy (white columns) noted in the
487 water-only tests with *B. sowerbyi* and *T. tubifex* exposed to a concentration series of As
488 and Zn for 96 hours. **Asterisks** indicate significant differences as compared to the
489 control treatment (Fisher's exact test; $p = 0.05$)

490 **Fig. 2.** Body appearance transformations observed in the tests with *T. tubifex* and *B.*
491 *sowerbyi*:

492 **a.** *T. tubifex* exposed for 14 days to Zn ($679 \text{ mg kg}^{-1} \text{ d.w.}$): swollen anterior body parts

493 **b.** *T. tubifex* exposed to Zn (1577 mg L^{-1}) for 96 hours: with swollen body and increased
494 number of granules in the coelomatic cavity (as indicated by the arrows)

495 **c.** *T. tubifex* after 96 hours exposure to 118 mg As L^{-1} : deformed body posterior
496 appearing the formation of a bifid tail

497 **d** and **e.** *B. sowerbyi* exposed to Zn (0.85 mg L^{-1} ; D) and As (52 mg L^{-1} ; E) with parts of
498 the posterior body lacking gills and with ring constrictions

499 Scale bars: A, D and E: 2 mm; B: 1 mm; C: 0.5 mm

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Table 1. Metal concentrations (mean \pm SE; in mg kg⁻¹ dry weight) and physical characteristics of the natural sediment used for the cultures and tests in Brazil (Perdizes), and culture (Iturbaz) and tests (Barrundia) in Spain with *B. sowerbyi* and *T. tubifex*, respectively.

	Perdizes	Iturbaz	Barrundia
Metal concentrations			
As	5.0 \pm 0.7	4.3 \pm 1.9	2.7 \pm 1.4
Cd	0.004 \pm 0.009	0.5 \pm 0.5	0.2 \pm 0.1
Co	0	3.7 \pm 4.7	8.9 \pm 0.3
Cr	0.030 \pm 0.035	13 \pm 5	22 \pm 25
Cu	0.006 \pm 0.009	2.4 \pm 1.0	3.3 \pm 1.6
Hg	ND	0.09 \pm 0.01	0.2 \pm 0.2
Ni	0.01 \pm 0.02	10 \pm 5	16 \pm 0.2
Pb	0.1 \pm 0.1	7.9 \pm 1.6	8.6 \pm 6.4
Zn	56 \pm 25	9.6 \pm 2.9	27 \pm 28
Physical characteristics of the sediment			
% OM	0.8 \pm 0.4	3.1 \pm 0.5	1.7 \pm 0.1
% SDW	78 \pm 0.3	ND	70 \pm 2
% grain size fractions	46 % 0.5-0.25 mm	3% gravel	49 % 0.5-0,25mm
	48 % 0.25-0.105 mm	68% sand	39 % 0.25-0.125mm
	5 % 0.105-0.53 mm	29% silt-clay	8% 0.125-0.063 mm
	1 % <0.053 mm		4% < 0.063 mm

OM = organic matter content; SDW = soil dry weight fraction; ND = not determined

Table 2. Nominal and measured test concentrations of arsenic and zinc in the water-only (96h; in mg L⁻¹), and sediment (in mg kg⁻¹ d.w.) and pore water (in mg L⁻¹) of the sediment (14d) toxicity tests performed with *Tubifex tubifex* and *Branchiura sowerbyi*.

Water-only 96h						
	As			Zn		
	Nominal	Measured	Recovery rate ^a	Nominal	Measured	Recovery rate ^a
<i>B. sowerbyi</i>	Control	0	-	Control	0	-
	8	6.9	86	0.62	0.50	80
	16	13	82	1.3	0.85	68
	32	27	85	2.5	2.0	79
	64	52	81	5	4.3	87
	128	81	63	10	8.9	89
<i>T. tubifex</i>	Control	0.03	-	Control	0.03	-
	4	4.0	100	1.25	1.34	107
	8	7.98	99	2.5	2.75	110
	16	15	96	5	5.5	110
	32	32	100	10	13	134
	64	66	103	20	21	106
	128	118	92	40	41	103
Sediment-water 14d						
	As ^b			Zn		
	Nominal	Sediment	Pore water	Nominal	Sediment	Pore water
<i>B. sowerbyi</i>	-	-	-	Control	80	0.03
	-	-	-	50	168	0.63
	-	-	-	87.5	230	1.9
	-	-	-	153	280	8.6
	-	-	-	268	358	15
	-	-	-	469	373	28
<i>T. tubifex</i>	Control	12	0.05	Control	181	0.009
	3.2	29	0.03	10	152	0.008
	10	44	0.12	31.6	259	0.009
	31.6	95	0.45	100	383	0.016
	100	208	3.84	316	565	0.43
	316	421	18.79	1000	679	9.57

^a – Recovery rate: ratio between measured and nominal concentration.

^b - The sediment 14d toxicity test for As with *B. sowerbyi* was not performed due to logistic constraints.

Table 3. Physical-chemical water quality parameters (mean \pm SD) in the water-only (96h) and sediment (14d) tests test evaluating the toxicity of arsenic and zinc to *Tubifex tubifex* and *Branchiura sowerbyi*.

Water-only 96h								
	As				Zn			
	pH (-)	DO (mg L ⁻¹)	EC (μ S cm ⁻¹)	Ammonia (mg L ⁻¹)	pH (-)	DO (mg L ⁻¹)	EC (μ S cm ⁻¹)	Ammonia (mg L ⁻¹)
<i>B. sowerbyi</i>	7.4 \pm 0.3	6.6 \pm 0.8	141 \pm 27	0.4 \pm 0.1	7.1 \pm 0.3	6.5 \pm 1.0	118 \pm 12	0.4 \pm 0.1
<i>T. tubifex</i>	7.7 \pm 0.2	7.7 \pm 0.6	290 \pm 35	0.5 \pm 0.1	7.1 \pm 0.1	7.7 \pm 0.5	293 \pm 33	0.4 \pm 0.2
Sediment-water 14d								
	As				Zn			
	pH (-)	DO (mg L ⁻¹)	EC (μ S cm ⁻¹)	Ammonia (mg L ⁻¹)	pH (-)	DO (mg L ⁻¹)	EC (μ S cm ⁻¹)	Ammonia (mg L ⁻¹)
<i>B. sowerbyi</i>	*	*	*	*	7.4 \pm 0.5	6.6 \pm 0.3	204 \pm 47	2.1 \pm 0.2
<i>T. tubifex</i>	7.9 \pm 0.3	7.6 \pm 0.4	310 \pm 41	1.5 \pm 0.2	8.0 \pm 0.5	7.7 \pm 0.3	325 \pm 49	1.9 \pm 0.2

* The sediment 14d toxicity test for As with *B. sowerbyi* was not performed due to logistic constraints.

Table 4 - Values of the Lethal Concentrations (LC₁₀ and LC₅₀) and Effect Concentrations (EC₁₀ and EC₅₀ for autotomy) for *Branchiura sowerbyi* and *Tubifex tubifex* exposed to As and Zn in the 96h water-only (in mg L⁻¹) and 14d sediment-water (in mg kg⁻¹d.w.) tests (value ± SE).

Test	Metal	Species	Endpoint	LC ₁₀ /EC ₁₀	LC ₅₀ /EC ₅₀	
96h	As	<i>B. sowerbyi</i>	mortality	15.35 ± 2.13	22 ± 1.8	
			autotomy	14.84 ± 1.93	21 ± 1.8	
		<i>T. tubifex</i>	mortality	74.60 ± 11.79	> 118.18	
	Zn	<i>B. sowerbyi</i>	mortality	0.85 ± 0.03	0.97 ± 0.07	
			autotomy	0.77 ± 0.34	0.88 ± 0.11	
		<i>T. tubifex</i>	mortality	4.56 ± 0.67	8.7 ± 0.84	
autotomy			4.57 ± 0.48	6.9 ± 0.59		
14d	As	<i>T. tubifex</i>	mortality	116.01 ± 34.70	251 ± 47	
			autotomy	80.98 ± 26.81	210 ± 44	
	Zn	<i>B. sowerbyi</i>	mortality	269.11 ± 46.14	280 ± 2.3	
			<i>T. tubifex</i>	mortality	675 ± 28.80	> 679
				autotomy	556 ± 38.59	635 ± 25

Figure 1

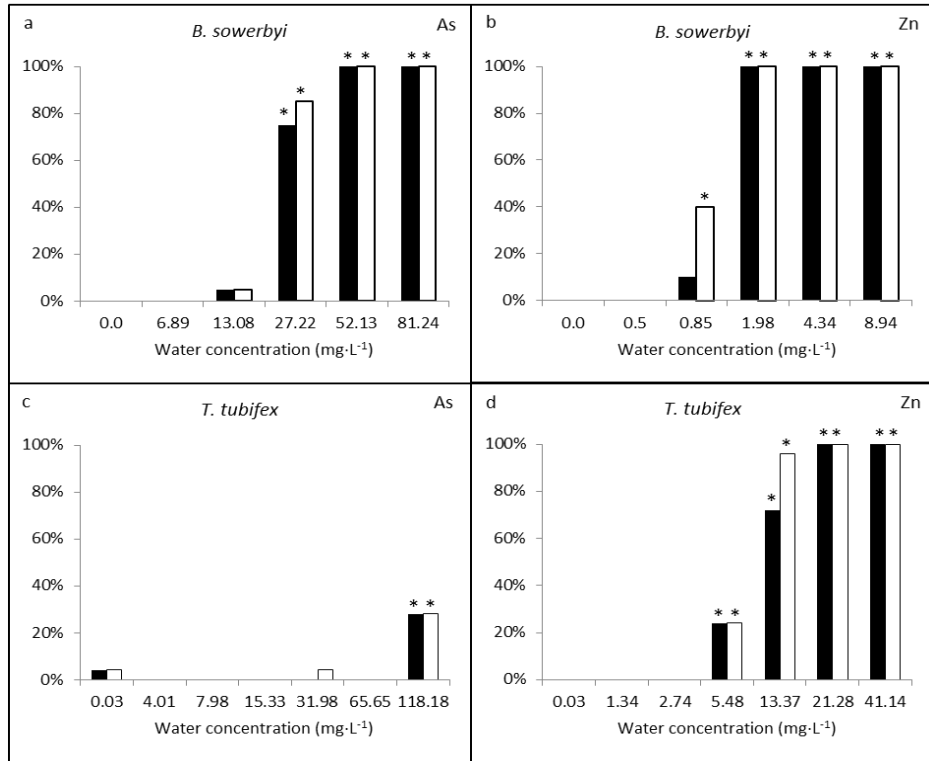


Figure 2

