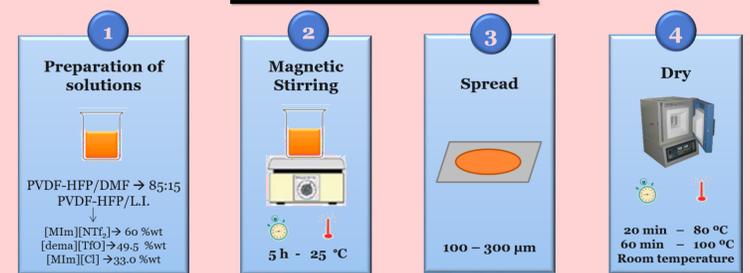


# Ionic liquids in the control of the poly(vinylidene fluoride-co-hexafluoropropylene) membranes morphology

## INTRODUCTION

The development of **polymer membranes** with **tailored micro-morphology** and wettability is a demand in the areas of filtration, sensors or tissue engineering, among others. Poly(vinylidene fluoride-co-hexafluoropropylene) (PVDF-HFP) is a thermoplastic copolymer and one of the most interesting polymers to be used in these areas due to its good properties. However, the control of the morphology is a complicated task and is mostly restricted to the use of solvent evaporation (SE) techniques [1]. In this way, ionic liquids (ILs), molten salts with melting points lower than 100 °C, which stand out for their good properties, such as high thermal stability or nonflammability, are a promising alternative for the **control of morphology** in certain materials since the large number of ionic liquids that exist allows to tailor the most suitable combination in order to meet the desired properties [2-3]. In this work, the production and the characterization of PVDF-HFP@ionic liquid **composite membranes** using different ionic liquids and methodologies are described in detail (figs. 1-6).

## EXPERIMENTAL



The composite properties were analyzed by **Scanning Electron Microscopy - SEM** (JEOL JSM-7000F), **Contact Angle** (NEURTEK OCA 15EC DATAPHYSICS), **Mechanical Test** (Shimadzu AGS-500NJ), **Thermogravimetric Analysis - TGA** (NETZSCH STA 449F3) and **Infrared Spectroscopy - FTIR** (Jasco FT/IR-6100 spectrometer).

## RESULTS

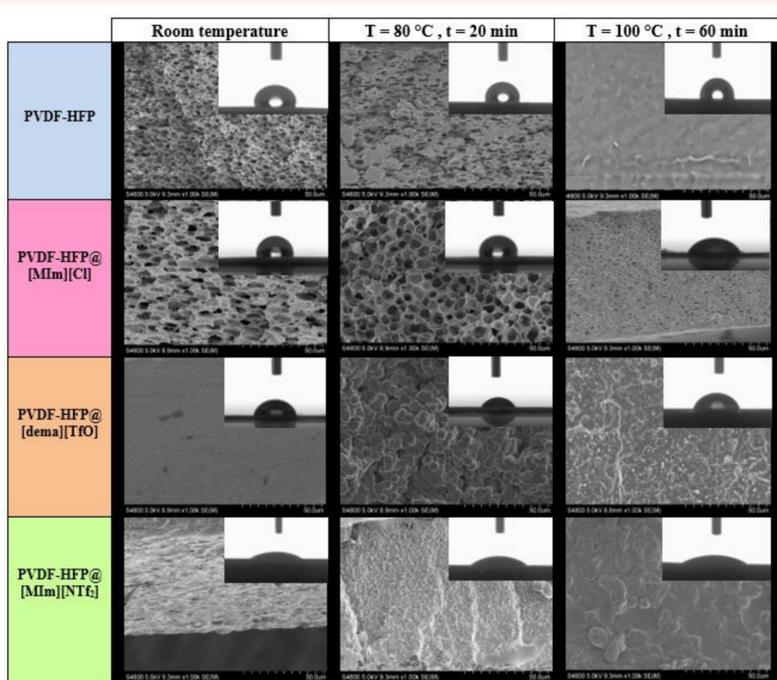


Figure 1 : SEM (cross-section) and contact angle images for the different samples.

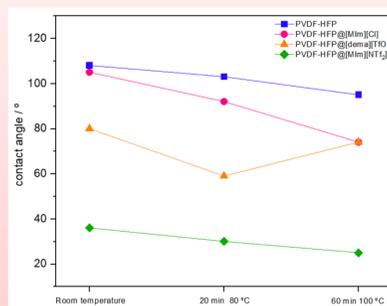


Figure 2 : Contact angle estimated for the samples.

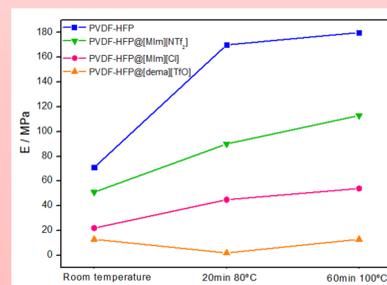
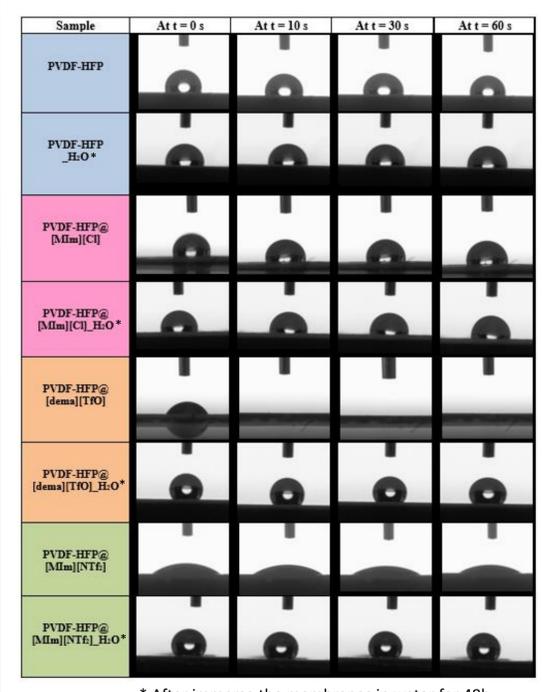


Figure 4 : Elastic module estimated for the samples.



\* After immerse the membranes in water for 48h.

Figure 3 : Images of the Contact Angle Measurements as a function of time of the different samples dried at 80 °C (before and after immersion in H<sub>2</sub>O).

Different morphologies and properties in function of the ionic liquid present in the membrane.

After immerse the membranes in water we can appreciate a **higher hydrophobicity** on the membranes, nonetheless, the membranes **morphology is not affected**. The percentage of polymer piezoelectric  $\beta$ -phase is maintained or even increased.

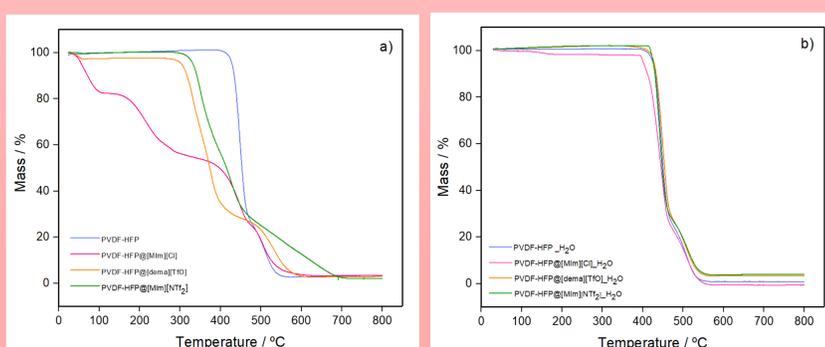


Figure 5 : Thermogravimetric curves for the membranes dried at 80 °C, (a) before immersion in water (b) after immersion in water.

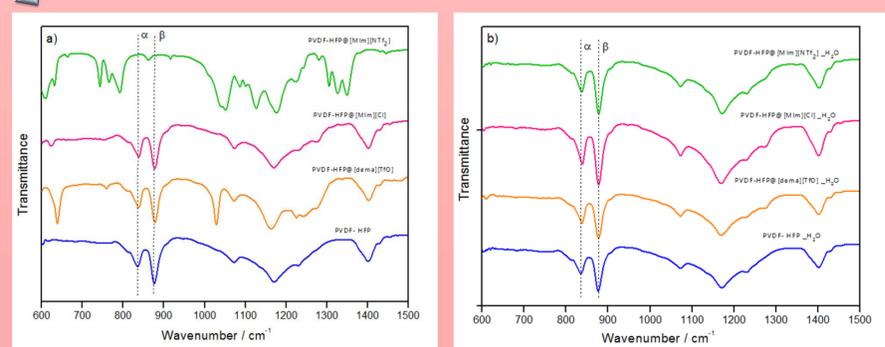


Figure 6 : FTIR-ATR spectra of the different membranes dried at 80 °C, (a) before immersion in water (b) after immersion in water.

## DISCUSSION & CONCLUSIONS

- The polymer membrane properties can be **tuned** using ILs.
- **Morphology, wettability** or **mechanical properties** change **depend** on the production methodology employed as well as on the type of **ionic liquid** used.
- After the immersion of membranes in water, the morphology is maintained, but it recovers its hydrophobic properties as well as its thermal stability until temperatures higher than 400 °C. The percentage of piezoelectric  $\beta$ -phase is maintained or even increased.
- The possibility to **tailor** the membranes **morphology** by the variation of the IL type opens **new possibilities** in the area of membranes production, since the wide range of different IL structures predicts a **huge variety** of different membrane structures.

## REFERENCES

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