

SUPPLEMENTARY MATERIAL

1. Analysis of the possible hydrogen bonds.

Table 1. Distances between the nitrogen atoms of the ammonium groups and the possible H-bond acceptor atoms.

Donor-Acceptor	Distance (Å)
N(1)-(4)	2.829(7)
N(1)-F(2) ⁱ	2.840(7)
N(1)-O(3) ⁱⁱ	2.870(5)
N(1)-O(2) ⁱⁱⁱ	2.997(6)
N(1)-O(1)	3.137(7)
N(1)-O(5) ⁱⁱ	3.172(7)
N(1)-F(1) ⁱ	3.225(6)
N(1)-O(6) ⁱⁱ	3.251(8)
N(1)-O(7) ⁱ	3.282(7)
N(2)-O(6)	2.760(6)
N(2)-O(5) ^{iv}	2.762(8)
N(2)-O(7) ^v	2.856(7)
N(2)-F(1) ^{vi}	2.925(7)
N(2)-O(3) ^{vii}	3.033(8)
N(2)-O(4) ^{iv}	3.119(8)
N(2)-F(2) ^{vii}	3.307(7)
N(2)-O(8) ^{iv}	3.433(7)
N(2)-O(1) ^{vi}	3.458(8)

ⁱ= $\frac{1}{2}-x, -\frac{1}{2}+y, -\frac{1}{2}+z$ ^v= $-1/2+x, 3/2-y, z$
ⁱⁱ= $-x, -y, -1/2+z$ ^{vi}= $-1/2+x, 1/2-y, z$
ⁱⁱⁱ= $x, -1+y, z$ ^{vii}= $-x, 1-y, -1/2+z$
^{iv}= $x, 1+y, z$

2. Impedance Spectroscopy Results.

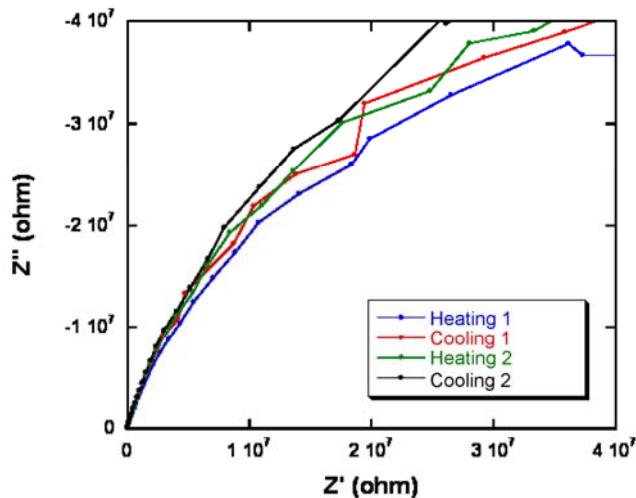


Fig. 1. Impedance spectrum of the heating and cooling processes at 220 °C.

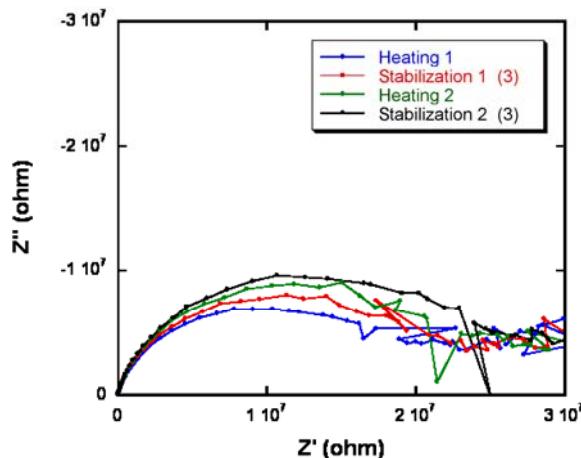


Fig. 2. Impedance spectrum of the heating and cooling processes at 260 °C.

Table 2.- Capacity and resistance values obtained from the fitting of the impedance spectra by means of simulation, with Zview software, at various temperatures.

Tempertura (°C)	C (F)	R (ohm)
260	$1.624 \cdot 10^{-11}$	$2.6498 \cdot 10^7$
240	$1.5336 \cdot 10^{-11}$	$5.5517 \cdot 10^7$
220	$1.7189 \cdot 10^{-11}$	$1.2323 \cdot 10^8$

Tabla 3.- Conductivity values obtained from the resistances at various temperature. Valores

Tempertura (°C)	σ (S/cm)
260	$8.734 \cdot 10^{-9}$
240	$4.169 \cdot 10^{-9}$
220	$1.878 \cdot 10^{-9}$